

APPENDIX 7. ESTIMATED MEAN OCEANIC CONCENTRATIONS OF THE ELEMENTS

Atomic number	Element	Species	Type of distribution ^a	Oceanic mean concentration (ng kg^{-3})
1	Hydrogen	H_2O	—	—
2	Helium	Dissolved gas	c	7.6
3	Lithium	Li^+	c	180×10^3
4	Beryllium	BeOH^+	s + n	0.21
5	Boron	$\text{B}(\text{OH})_3$	c	4.5×10^6
6	Carbon	Inorganic ΣCO_2	n	27.0×10^6
7	Nitrogen	Dissolved N_2	c	8.3×10^6
		NO_3^-	n	0.42×10^6
8	Oxygen	Dissolved O_2	inverse n	2.8×10^6
9	Fluorine	F^-	c	1.3×10^6
10	Neon	Dissolved gas	c	160
11	Sodium	Na^+	c	10.78×10^9
12	Magnesium	Mg^{2+}	c	1.28×10^9
13	Aluminum	$\text{Al}(\text{OH})_3^0$	s	30
14	Silicon	H_4SiO_4^0	n	2.8×10^6
15	Phosphorus	NaHPO_4^-	n	62×10^3
16	Sulfur	SO_4^{2-}	c	898×10^6
17	Chlorine	Cl^-	c	19.35×10^9
18	Argon	Dissolved gas	c	0.62×10^6
19	Potassium	K^+	c	399×10^6
20	Calcium	Ca^{2+}	almost c	412×10^6
21	Scandium	$\text{Sc}(\text{OH})_3^0$	(s + n)	0.7
22	Titanium	$\text{Ti}(\text{OH})_4^0$	s + n	6.5
23	Vanadium	NaHVO_4^-	almost c	2.0×10^3
24	Chromium	$\text{CrO}_4^{2-}(\text{VI})$	r + n	210
		$\text{Cr}(\text{OH})_3^0 (\text{III})$	r + s	2
25	Manganese	Mn^{2+}	s	20
26	Iron	$\text{Fe}(\text{OH})_3^0$	s + n	30
27	Cobalt	$\text{Co}(\text{OH})_2^0?$	s	1.2
28	Nickel	Ni^{2+}	n	480
29	Copper	CuCO_3^0	s + n	150
30	Zinc	Zn^{2+}	n	350
31	Callium	$\text{Ga}(\text{OH})_4^-$	s + n	1.2
32	Germanium	H_4GeO_4^0	n	5.5
33	Arsenic	$\text{HAsO}_4^{2-}(\text{V})$	r + n	1.2×10^3
		$\text{As}(\text{OH})_3^0(\text{III})$	r + s	5.2
34	Selenium	$\text{SeO}_4^{2-}(\text{VI})$	r + n	100
		$\text{SeO}_3^{2-}(\text{IV})$	r + n	55
35	Bromine	Br^-	c	67×10^6
36	Krypton	Dissolved gas	c	310
37	Rubidium	Rb^+	c	0.12×10^6
38	Strontium	Sr^{2+}	almost c	7.8×10^6
39	Yttrium	YCO_3^+	n	17
40	Zirconium	$\text{Zr}(\text{OH})_5^-$	s + n	15
41	Niobium	$\text{Nb}(\text{OH})_6^-$	almost c	0.37
42	Molybdenum	MoO_4^{2-}	c	10×10^3
43	Technetium	TcO_4^-	—	—
44	Ruthenium	RuO_4^-	?	< 0.005
45	Rhodium	$\text{Rh}(\text{OH})_3^0?$	n	0.08
46	Palladium	$\text{PdCl}_4^{2-}?$	n	0.06
47	Silver	AgCl_2^-	n	2
48	Cadmium	CdCl_2^0	n	70
49	Indium	$\text{In}(\text{OH})_3^0$	s	0.01
50	Tin	$\text{SnO}(\text{OH})_3^-$	s	0.5

Appendix 7 *Continued*

Atomic number	Element	Species	Type of distribution ^a	Oceanic mean concentration (ng kg^{-1})
51	Antimony	Sb(OH)_6^-	s?	200
52	Tellurium	Te(OH)_6^0	r + s	0.05
		TeO(OH)_3^-	r + s	0.02
53	Iodine	IO_3^-	almost c	58×10^3
		I^-	(r + s)	4.4
54	Xenon	Dissolved gas	c	66
55	Cesium	Cs^+	c	306
56	Barium	Ba^{2+}	n	15×10^3
57	Lanthanum	LaCO_3^+	n	5.6
58	Cerium	Ce(OH)_4^0	s	0.7
59	Praseodymium	PrCO_3^+	n	0.7
60	Neodymium	NdCO_3^+	n	3.3
61	Promethium	—	—	—
62	Samarium	SmCO_3^+	n	0.57
63	Europium	EuCO_3^+	n	0.17
64	Gadolinium	GdCO_3^+	n	0.9
65	Terbium	TbCO_3^+	n	0.17
66	Dysprosium	DyCO_3^+	n	1.1
67	Holmium	HoCO_3^+	n	0.36
68	Erbium	ErCO_3^+	n	1.2
69	Thulium	TmCO_3^+	n	0.2
70	Ytterbium	YbCO_3^+	n	1.2
71	Lutetium	LuCO_3^+	n	0.23
72	Hafnium	Hf(OH)_5^-	s + n	0.07
73	Tantalum	Ta(OH)_5^0	s + n	0.03
74	Tungsten	WO_4^{2-}	c	10
75	Rhenium	ReO_4^-	c	7.8
76	Osmium	OsO_4^0	almost c	0.009
77	Iridium	Ir(OH)_3^0	s?	0.00013
78	Platinum	PtCl_4^{2-}	c	0.05
79	Gold	$\text{AuOH(H}_2\text{O)}^0$	c	0.02
80	Mercury	HgCl_4^{2-}	(s + n)	0.14
81	Thallium	TI^+	c	13
82	Lead	PbCO_3^0	anth. + s	2.7
83	Bismuth	Bi(OH)_3^0	s	0.03
84	Polonium	PoO(OH)_3^-	s	—
85	Astatine	—	—	—
86	Radon	Dissolved gas	c	—
87	Francium	Fr^+	—	—
88	Radium	Ra^{2+}	n	0.00013
89	Actinium	AcCO_3^+	n	—
90	Thorium	Th(OH)_4^0	s	0.02
91	Protactinium	$\text{PaO}_2(\text{OH})^0$	s	—
92	Uranium	$\text{UO}_2(\text{CO}_3)_3^{4-}$	c	3.2×10^3
93	Neptunium	NpO_2^+	—	—
94	Plutonium	$\text{PuO}_2(\text{CO}_3)(\text{OH})^-$	(r + s)	—
95	Americium	AmCO_3^+	(s + n)	—

^ac, conservative; n, nutrient-like; s, scavenged; r, redox sensitive; anth., anthropogenic.

Vertical Profiles of Elements in the North Pacific Ocean

(compiled by Y. Nozaki, 2001)

